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The Crystal Structure of Nickel(II) Bis(dithiocarbamate)

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THOUGH the structures of several metal complexes of substituted dithiocarbamic acids have been studied,¹ nothing has yet been done with unsubstituted dithiocarbamates. We have now prepared the following compounds and studied them by X-ray methods, using Cu-K α radiation:

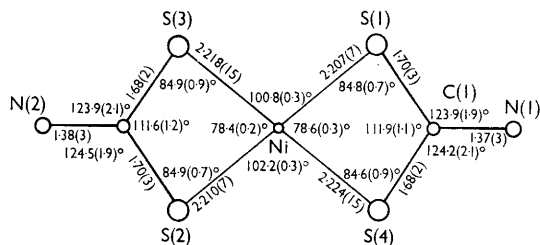
Ammonium dithiocarbamate, NH₂CS₂NH₄, $M = 110.2$. Orthorhombic, $a = 5.639$ (4), $b = 8.272$ (2), $c = 10.663$ (3) Å, $U = 497.4$ Å³, $D_m = 1.46$, $Z = 4$, $D_c = 1.471$; space group, $Pbcm$ or $Pbc2_1$.

Cobalt(III) tris(dithiocarbamate), (NH₂CS₂)₃Co^{III}, $M = 335.4$. Monoclinic, $a = 7.13$ (3), $b = 10.11$ (1), $c = 16.32$ (2) Å, $\beta = 100.8^\circ$ (7), $U = 1156$ Å³, $D_m = 1.96$, $Z = 4$, $D_c = 1.926$; space group, $P2_1/c$.

Nickel(II) bis(dithiocarbamate), (NH₂CS₂)₂Ni^{II}, $M = 243.0$. Monoclinic, $a = 8.96$ (2), $b = 9.825$ (5), $c = 11.28$ (2) Å, $\beta = 129.3^\circ$ (1), $U = 769.0$ Å³, $D_m = 2.09$, $Z = 4$, $D_c = 2.098$; space group, $P2_1/c$.

The structure of this last has been determined and refined by differential syntheses based on 1292 independent, photographically-determined intensities, leading to an R -value of 11.2%. Bond-distances and -angles are shown in the Figure; they are not significantly different from those

found in the α -form of nickel(II) bis-(*NN*-diethyldithiocarbamate).² If bond-distances only are considered, the whole molecule is practically centrosymmetrical about the metal atom; but this symmetry is not complete, since the four S-atoms and the Ni-atom deviate slightly but significantly from coplanarity. The four (heavy) atoms of each ligand molecule are coplanar.



FIGURE

Apart from the four S-atoms two other atoms make contacts with the Ni: these are an H-atom at 2.85 Å and an S-atom at 3.60 Å. Metal-hydrogen contacts of this type have already been noted.³

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¹ G. Peyronel, *Z. Krist.*, 1941, **103**, 157.

² M. Bonamico, G. Dessy, C. Mariani, A. Vaciago, and L. Zambonelli, *Acta Cryst.*, 1965, **19**, 619.

³ M. Bonamico, G. Dessy, A. Mugnoli, A. Vaciago, and L. Zambonelli, *Acta Cryst.*, 1965, **19**, 886; N. A. Bailey, J. M. Jenkins, R. Mason, and B. L. Shaw, *Chem. Comm.*, 1965, 237; J. A. Ibers, *Abstr. Amer. Cryst. Assoc.*, 1965, B 10.